

Synthesis of Quinolines through Three-Component Cascade Annulation of Aryl Diazonium Salts, Nitriles, and Alkynes

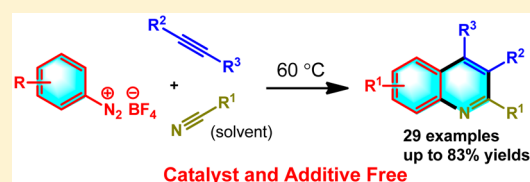
Hao Wang,[†] Qian Xu,[†] Sheng Shen,^{*,‡} and Shouyun Yu^{*,†}

[†]State Key Laboratory of Analytical Chemistry for Life Science, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing 210023, China

[‡]Department of Sports Medicine and Adult Reconstructive Surgery, Drum Tower Hospital, School of Medicine, Nanjing University, Nanjing 210008, China

S Supporting Information

ABSTRACT: An efficient and rapid synthesis of multiply substituted quinolines is described. This method is enabled by a three-component cascade annulation of readily available aryl diazonium salts, nitriles, and alkynes. This reaction is catalyst- and additive-free. Various aryl diazonium salts, nitriles, and alkynes can participate in this transformation, and the yields are up to 83%.



Quinolines represent a group of ubiquitous heterocycles in the natural alkaloids displaying a wide variety of pharmacological properties, such as tumoricidal, angina pectoris, antihypertensive, and antibacterial activities.¹ Furthermore, their structures and properties have made them widely applicable in functional materials² and asymmetric catalysts.³ The synthesis of the structural core of quinolines has been well established.⁴ The most popular strategy for the construction of quinoline core is based on the condensation of aniline derivatives with various carbonyl compounds,^{4,5} e.g., Combes synthesis, Conrad–Limpach–Knorr synthesis, and Friedlander synthesis. Recently, transition-metal-catalyzed couplings have emerged as a powerful tool for the efficient and rapid synthesis of quinolines.^{4b,6} Despite these advances, a clean, efficient, and economic technology is still needed to obtain useful polyfunctionalized quinolines from readily available starting materials.

Cycloaddition of *N*-arylnitrilium salts (I) with alkynes can compete with classical syntheses in the efficacy and rapidity of the quinoline construction (Figure 1). Compared to well-established annulation of iminium salts with alkynes followed by oxidation to quinolines,⁷ annulation of nitrilium salts with alkynes can give quinolines directly without oxidation. In 1992, Jochims and co-workers reported that *N*-arylnitrilium salts (I) could be prepared by Beckmann rearrangement of the corresponding oximes (II) or abstraction of chloride from the corresponding imidoyl chlorides (III).⁸ The preformed nitrilium salts (I) could undergo polar [4 + 2] cycloaddition with alkynes to give substituted quinolines (Figure 1A). Recently, Chen and co-workers found that *N*-arylnitrilium salts (I) could be generated *in situ* by copper-catalyzed aryl group transfer from diaryliodonium salts (IV) to nitriles, followed by annulation with alkynes to give quinolines (Figure 1B).^{6a}

Encouraged by these elegant reports, we sought to make a modified approach for the synthesis of quinolines via cascade

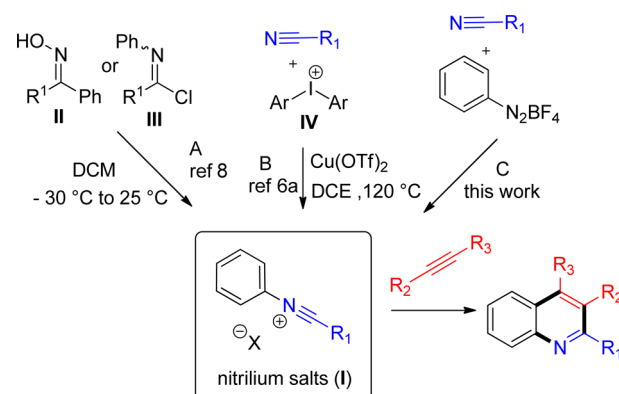


Figure 1. Syntheses of quinolines by cycloaddition of *N*-arylnitrilium salts with alkynes.

annulation of aryl diazonium salts, nitriles, and alkynes.⁹ It is known that diazonium salts decompose on warming into an aryl cation with the release of N₂. The inherent electrophilicity of diazonium salts offers a pathway to introduce a cyano group into an aromatic ring to yield *N*-arylnitrilium salts (I). As demonstrated by Jochims and Chen, these *N*-arylnitrilium salts can undergo [4 + 2] cycloaddition with alkynes to provide polysubstituted quinolines (Figure 1C). Herein, we would like to report a rapid and efficient method to synthesize quinolines from three easily accessible precursors, aryl diazonium salts, nitriles, and alkynes. This method presented here is catalyst- and additive-free, economic, and environmentally benign.

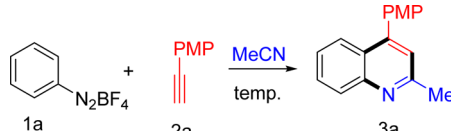
At the outset of this investigation, we explored the cascade reaction of aryl diazonium salt **1a** and *p*-methoxyphenylacetylene (**2a**) using acetonitrile as the solvent. We were pleased to find that when a mixture of aryl diazonium salt **1a**

Received: October 16, 2016

Published: December 12, 2016

(0.3 mmol) and *p*-methoxyphenylacetylene (**2a**, 0.6 mmol) in 3 mL of anhydrous acetonitrile was stirred at room temperature for 30 h, the desired quinoline **3a** was obtained in a 55% isolated yield (Table 1, entry 1). When the reaction mixture

Table 1. Optimization of the Reaction Conditions^a



entry	MeCN	temp	time	yield/% ^b
1	3 mL	25 °C	30 h	55
2	3 mL	40 °C	6 h	82
3	1 mL	40 °C	6 h	76
4	2 mL	40 °C	6 h	80
5	6 mL	40 °C	6 h	72
6	9 mL	40 °C	6 h	66
7	2 mL	60 °C	1 h	84
8	2 mL	80 °C	0.5 h	83
9 ^c	2 mL	40 °C	6 h	72
10 ^d	2 mL	40 °C	6 h	74
11 ^c	2 mL	60 °C	1 h	79
12 ^d	2 mL	60 °C	1 h	83

^aReaction conditions: A solution of **1a** (0.3 mmol), **2a** (0.6 mmol) in anhydrous acetonitrile was stirred at the indicated temperature. ^bIsolated yields. ^c1.2 equiv of **2a** (0.36 mmol) was used. ^d1.5 equiv of **2a** (0.45 mmol) was used. PMP = *para*-methoxyphenyl.

was slightly warmed up to 40 °C, the reaction reached completion in 6 h with 82% isolated yield (entry 2). The concentration of the reaction mixture had a significant impact on the chemical yields (entries 3–6). A more concentrated or diluted reaction mixture led to lower isolated yields. A higher temperature could accelerate the reaction with comparable yields (entries 7–8). The reaction could finish in 1 h at 60 °C and 0.5 h at 80 °C. The dosage of alkyne **2a** can be reduced to 1.2 and 1.5 equiv with 79% and 83% isolated yields respectively at 60 °C (entries 11–12).

After the reaction parameter was established, we next explored the scope and limitation of this reaction (Scheme 1). First, a variety of alkynes reacted with diazonium salt **1a** in acetonitrile. Electron-rich and -neutral phenylacetylene derivatives underwent reaction smoothly with 68–83% yields while electron-deficient phenylacetylene derivatives were less reactive with 37–54% yields. Naphthalene- and thiophene-derived terminal alkynes could also undergo this transformation frequently to provide the quinolines **3k** and **3l** in 78% and 77% yields, respectively. To our delight, internal aromatic alkynes were also suitable reaction partners to give the corresponding quinolines **3m–3o** in reasonable isolated yields (53–73%). The reaction with linear aliphatic alkyne (1-octyne) was sluggish, and only 33% desired quinolone **3p** could be isolated even though 3.0 equiv of 1-octyne were employed. However, ethynylcyclopropane was a good reaction partner and a 73% yield of quinoline **3q** could be obtained under optimal conditions. Then, various functionalized aryl diazonium salts reacted with *para*-methoxyphenylacetylene (**2a**) in acetonitrile. Functional groups, such as ester (**3s**), ether (**3w**), trifluoromethyl (**3t**), ketone (**3u**), and naphthalene (**3x**), could be tolerated. When *meta*-methoxyphenyl diazonium salt was used, two regioisomers **3w** and **3w'** were obtained with a 77%

combined yield. Annulation of phenyl diazonium salt (**1a**) and *para*-methoxyphenylacetylene (**2a**) in benzonitrile and pivalonitrile was also feasible. The corresponding quinolines **3y** and **3z** were isolated in 54% and 52% yields, respectively. Finally, multiple functionalized quinolines **3aa–3ac** could be also prepared in acceptable yields by utilizing this method. In the cases of low yields, the major byproducts were the amides, which were generated from the hydrolysis of the corresponding nitrilium intermediates.

In order to gain insights into the mechanism of this transformation, some control experiments were carried out (Scheme 2). When 10 equiv of distilled water were introduced into the model reaction mixture, no quinoline **3a** was observed. Instead, the *N*-phenylacetamide (**4**) was obtained in a 61% yield based on the crude ¹H NMR analysis. Furthermore, when alkyne **2a** was removed from the reaction mixture, a similar result was obtained. These findings suggests that the nitrilium salt is the key intermediate of this reaction.^{6a,10}

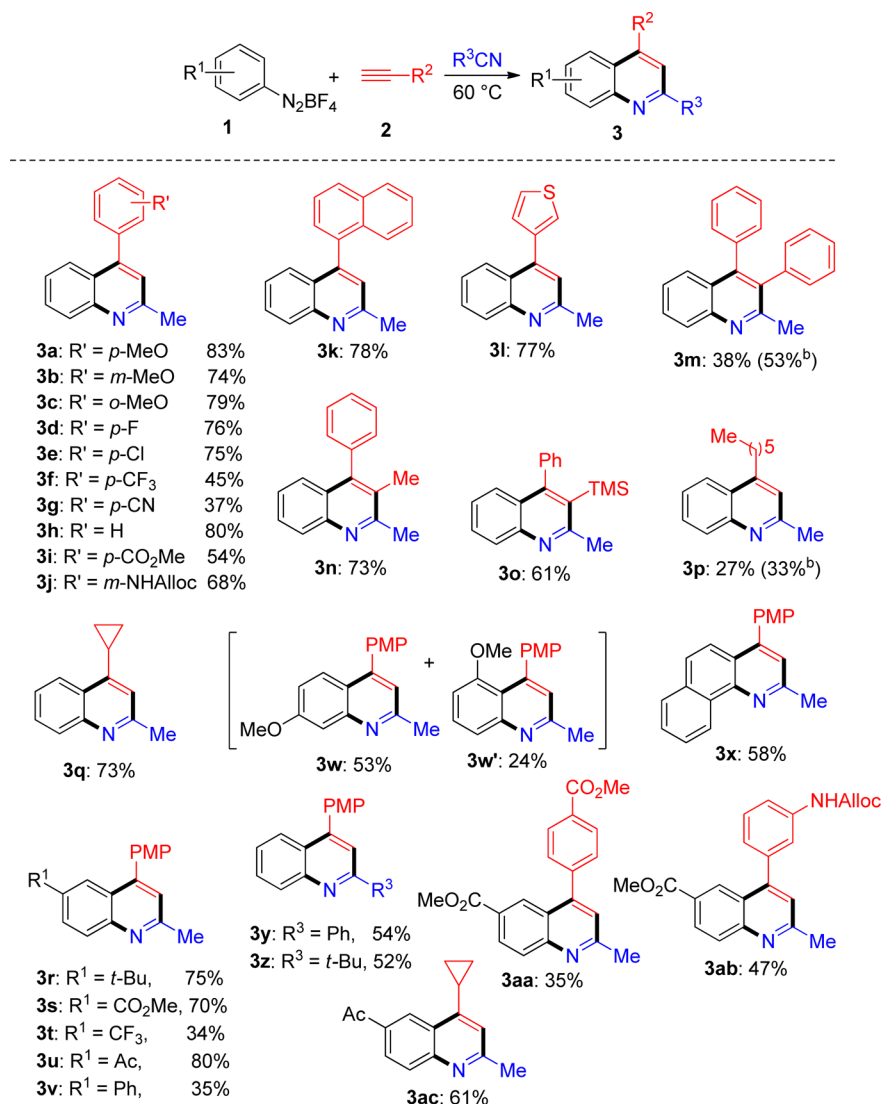
On the basis of control experiments and the previous reports,^{6a,8} a plausible mechanism is proposed (Figure 2). Upon heating, the aryl diazonium salt **1a** decomposes into the aryl cation **5** with concomitant release of N₂. Acetonitrile works as a nucleophile to attack the aryl cation **5**, resulting in the formation of nitrilium cation **6**. There are two possible pathways to form quinoline **3a**. In path A, a nitrilium cation reacts with alkyne **2a** through a concerted Diels–Alder reaction to give the intermediate **8**. After deprotonation, the intermediate **8** can convert to quinoline **3a**. Alternatively, quinoline **3a** can be generated in a stepwise manner (path B). The intermediate **6** is attacked by alkyne **2a** to give the vinyl cation **9** following by Friedel–Crafts-type cyclization to give the desired quinoline **3a**.

In summary, we have described an efficient and rapid synthesis of multiply functionalized quinolines from three components, aryl diazonium salts, alkynes, and nitriles. The reaction can finish in 1 h at 60 °C, and the yield afforded was up to 83%. Furthermore, neither a catalyst nor an additive is necessary. This economic and environmentally benign procedure should benefit the future development of potential industrial processes and makes this protocol particularly attractive for the chemical community.

EXPERIMENTAL SECTION

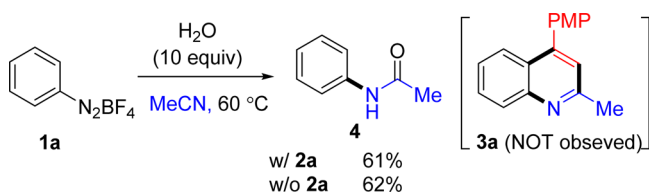
General Information. All reagents were used without further purification. Thin layer chromatography (TLC) was performed on precoated plates (silica gel 60 F254, Art 5715) and visualized by fluorescence quenching under UV light and by staining with phosphomolybdic acid or potassium permanganate, respectively. Column chromatography was performed on Silica Gel 60 (300–400 Mesh) using a forced flow of 0.5–1.0 bar. ¹H NMR (400 MHz), ¹³C NMR (101 MHz), and ¹⁹F (376 MHz) were measured on a 400 M NMR spectrometer. Chemical shifts are expressed in parts per million (ppm) with respect to the residual solvent peak. Coupling constants are reported as hertz (Hz), and signal shapes and splitting patterns are indicated as follows: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet. Infrared (IR) spectra were recorded on an IR spectrophotometer and are reported as wavenumber (cm⁻¹).

General Procedure for the Preparation of Aryl Diazonium Tetrafluoroborates.¹¹ The appropriate aniline (10 mmol) was dissolved in a mixture of 3.4 mL of hydrofluoroboric acid (50%) and 4 mL of distilled water. The reaction mixture was cooled down to 0 °C using an ice–water bath, and then a sodium nitrite (NaNO₂) solution (0.69 g in 1.5 mL) was added dropwise. The resulting reaction mixture was stirred for 40 min at 0–5 °C, and the obtained precipitate was collected by filtration, dried, and redissolved in a minimum amount of

Scheme 1. Substrate Scope^a

^aReaction conditions: A solution of 1 (0.3 mmol), 2 (0.45 mmol) in anhydrous nitrile (2 mL) was stirred at $60^\circ C$ for 1 h. ^b3.0 equiv of 2 were used.

Scheme 2. Control Experiments



acetone. Diethyl ether was added until precipitation of diazonium tetrafluoroborate, which is filtered, washed several times with small portions of diethyl ether, and dried under vacuum.

General Procedure for Synthesis of Quinolines. A 10 mL round-bottom flask equipped with a rubber septum and magnetic stir bar was charged with aryl diazonium salt 1a (0.3 mmol, 1.0 equiv). Then *p*-MeO phenylacetylene 2a (0.45 mmol, 1.5 equiv) and CH₃CN (2.0 mL) were added with a syringe. The mixture was placed in an oil bath preheated to $60^\circ C$. After the reaction was complete (as judged by TLC analysis), the mixture was poured into a separatory funnel containing 20 mL of H₂O and 20 mL of Et₂O. The layers were separated, and the aqueous layer was extracted with Et₂O (2 × 20 mL). The combined organic layers were dried over Na₂SO₄ and concentrated under reduced pressure after filtration. The crude

product was purified by flash chromatography on silica gel to afford the desired product 3a.

4-(4-Methoxyphenyl)-2-methylquinoline (3a):¹² Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 3a as a white solid (62.1 mg, 83%); ¹H NMR (400 MHz, CDCl₃) δ 8.08 (d, *J* = 8.4 Hz, 1H), 7.90 (dd, *J* = 8.4, 0.8 Hz, 1H), 7.67 (ddd, *J* = 8.3, 6.9, 1.4 Hz, 1H), 7.45–7.41 (m, 3H), 7.21 (s, 1H), 7.05 (d, *J* = 8.7 Hz, 2H), 3.89 (s, 3H), 2.76 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.8, 158.5, 148.5, 148.3, 130.8, 130.4, 129.3, 129.0, 125.7, 125.6, 125.3, 122.2, 114.0, 55.4, 25.4.

4-(3-Methoxyphenyl)-2-methylquinoline (3b). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 3b as a yellow solid (55.6 mg, 74%). Mp 72 – $73^\circ C$. IR (film, cm⁻¹): 1581; 1464; 1038; 887; 766 cm⁻¹. ¹H NMR (400 MHz, CDCl₃) δ 8.09 (d, *J* = 8.3 Hz, 1H), 7.88 (dd, *J* = 8.4, 0.8 Hz, 1H), 7.68 (ddd, *J* = 8.3, 6.9, 1.3 Hz, 1H), 7.44–7.40 (m, 2H), 7.23 (s, 1H), 7.07 (d, *J* = 7.7 Hz, 1H), 7.03–7.00 (m, 2H), 3.85 (s, 3H), 2.77 (s, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 159.6, 158.5, 148.41, 148.35, 139.5, 129.6, 129.4, 129.0, 125.8, 125.7, 125.1, 122.1, 121.9, 115.1, 113.8, 55.4, 25.4. HRMS (DART-FTICl Positive) ([M + H]⁺) calcd for C₁₇H₁₅NO: 250.1226; found: 250.1226.

4-(2-Methoxyphenyl)-2-methylquinoline (3c). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded 3c as a yellow solid (59.3 mg, 79%). Mp 105 – $106^\circ C$. IR (film, cm⁻¹): 1592;

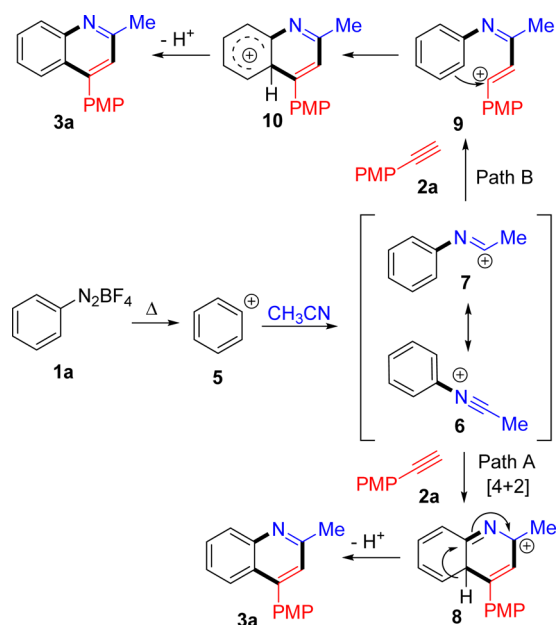


Figure 2. Proposed mechanism.

1487; 1023; 873; 763 cm^{-1} . $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.07 (d, $J = 8.4$ Hz, 1H), 7.64 (ddd, $J = 8.3, 6.9, 1.4$ Hz, 1H), 7.55–7.53 (m, 1H), 7.45 (td, $J = 8.3, 1.7$ Hz, 1H), 7.38–7.34 (m, 1H), 7.24 (dd, $J = 7.5, 1.8$ Hz, 1H), 7.22 (s, 1H), 7.08 (dt, $J = 8.3, 4.1$ Hz, 1H), 7.04 (d, $J = 8.3$ Hz, 1H), 3.69 (s, 3H), 2.76 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.5, 156.8, 148.0, 146.0, 131.2, 129.9, 129.1, 128.8, 127.0, 126.1, 125.8, 125.4, 123.0, 120.7, 111.1, 55.5, 25.4. HRMS (DART-FTICl Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{17}\text{H}_{15}\text{NO}$: 250.1226; found: 250.1225.

4-(4-Fluorophenyl)-2-methylquinoline (3d):¹² Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3d** as a yellow solid (54.3 mg, 76%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.09 (d, $J = 8.3$ Hz, 1H), 7.80 (d, $J = 8.4$ Hz, 1H), 7.68 (ddd, $J = 8.3, 6.9, 1.3$ Hz, 1H), 7.47–7.41 (m, 3H), 7.23–7.18 (m, 3H), 2.77 (s, 3H). $^{19}\text{F NMR}$ (377 MHz, CDCl_3) δ -113.40. $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 162.7 (d, $J = 248.0$ Hz), 158.5, 148.4, 147.5, 134.1, 134.1, 131.2 (d, $J = 8.2$ Hz), 129.3 (d, $J = 30.9$ Hz), 125.9, 125.36, 125.1, 122.3, 115.6 (d, $J = 21.6$ Hz), 25.3.

4-(4-Chlorophenyl)-2-methylquinoline (3e):¹³ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3e** as a yellow solid (56.7 mg, 75%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.09 (d, $J = 8.4$ Hz, 1H), 7.78 (d, $J = 8.3$ Hz, 1H), 7.70–7.66 (m, 1H), 7.49–7.40 (m, 5H), 7.19 (s, 1H), 2.77 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.5, 148.3, 147.3, 136.5, 134.6, 130.8, 129.5, 129.1, 128.8, 126.0, 125.3, 124.8, 122.2, 25.3.

2-Methyl-4-(4-(trifluoromethyl)phenyl)quinoline (3f):¹² Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3f** as a yellow solid (38.4 mg, 45%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.11 (d, $J = 8.4$ Hz, 1H), 7.80–7.69 (m, 4H), 7.61 (d, $J = 7.7$ Hz, 2H), 7.41–7.43 (m, 1H), 7.22 (s, 1H), 2.79 (s, 3H). $^{19}\text{F NMR}$ (377 MHz, CDCl_3) δ -62.57. $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.5, 148.4, 147.0, 141.8, 130.6 (q, $J = 32.6$ Hz), 129.9, 129.6, 129.2, 128.2, 126.2, 125.5 (q, $J = 3.6$ Hz), 124.6, 124.1 (q, $J = 272.3$ Hz), 122.2, 25.3.

4-(2-Methylquinolin-4-yl)benzotrile (3g):¹² Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3g** as a yellow solid (27.0 mg, 37%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.11 (d, $J = 8.3$ Hz, 1H), 7.83 (d, $J = 8.3$ Hz, 2H), 7.74–7.70 (m, 2H), 7.62 (d, $J = 8.3$ Hz, 2H), 7.49–7.45 (m, 1H), 7.22 (s, 1H), 2.79 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.6, 148.4, 146.4, 142.9, 132.4, 130.3, 129.8, 129.3, 126.4, 124.8, 124.3, 122.1, 118.5, 112.4, 25.4.

2-Methyl-4-phenylquinoline (3h):¹² Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3h** as a yellow solid (52.6 mg, 80%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.09 (d, $J = 8.4$ Hz, 1H), 7.84 (d, $J = 8.4$ Hz, 1H), 7.69–7.64 (m, 1H), 7.51–7.45 (m,

5H), 7.42–7.38 (m, 1H), 7.21 (s, 1H), 2.76 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.5, 148.6, 148.4, 138.2, 129.5, 129.3, 129.0, 128.5, 128.3, 125.8, 125.7, 125.1, 122.2, 25.4.

Methyl 4-(2-Methylquinolin-4-yl)benzoate (3i): Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3i** as a yellow solid (44.6 mg, 54%). Mp 110–111 $^\circ\text{C}$. IR (film, cm^{-1}): 1700; 1432; 1019; 858; 758 cm^{-1} . $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.19 (d, $J = 8.4$ Hz, 2H), 8.10 (d, $J = 8.3$ Hz, 1H), 7.77 (d, $J = 8.4$ Hz, 1H), 7.70 (ddd, $J = 8.3, 6.9, 1.3$ Hz, 1H), 7.57 (d, $J = 8.4$ Hz, 2H), 7.44 (ddd, $J = 8.2, 7.0, 1.1$ Hz, 1H), 7.23 (s, 1H), 3.98 (s, 3H), 2.79 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 166.7, 158.5, 148.4, 147.4, 142.8, 130.1, 129.8, 129.6, 129.5, 129.2, 126.1, 125.2, 124.6, 122.1, 52.3, 25.4. HRMS (DART-FTICl Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{18}\text{H}_{15}\text{NO}_2$: 278.1176; found: 278.1173.

Allyl (3-(2-Methylquinolin-4-yl)phenyl)carbamate (3j): Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3j** as a white solid (64.9 mg, 68%). Mp 166–167 $^\circ\text{C}$. IR (film, cm^{-1}): 1717; 1432; 1104; 858; 775 cm^{-1} . $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.07 (d, $J = 8.2$ Hz, 1H), 7.84 (dd, $J = 8.4, 0.8$ Hz, 1H), 7.65 (ddd, $J = 8.3, 6.9, 1.3$ Hz, 1H), 7.55–7.51 (m, 3H), 7.42–7.37 (m, 2H), 7.19 (s, 1H), 7.15 (d, $J = 7.5$ Hz, 1H), 5.94 (ddt, $J = 17.0, 10.5, 5.7$ Hz, 1H), 5.33 (dq, $J = 17.2, 1.5$ Hz, 1H), 5.23 (dd, $J = 10.4, 1.3$ Hz, 1H), 4.66 (d, $J = 5.7$ Hz, 2H), 2.74 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.5, 153.5, 148.3, 148.2, 139.0, 138.4, 132.4, 129.4, 129.2, 128.9, 125.8, 125.6, 125.0, 124.6, 122.2, 119.7, 118.6, 118.2, 65.9, 25.2. HRMS (DART-FTICl Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{20}\text{H}_{18}\text{N}_2\text{O}_2$: 319.1441; found: 319.1441.

2-Methyl-4-(naphthalen-1-yl)quinoline (3k): Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3k** as a white solid (63.0 mg, 78%). Mp 164–165 $^\circ\text{C}$. IR (film, cm^{-1}): 1597; 1505; 1020; 764; 739 cm^{-1} . $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.13 (d, $J = 8.4$ Hz, 1H), 7.94 (dd, $J = 11.0, 8.4$ Hz, 2H), 7.67–7.63 (m, 1H), 7.58–7.54 (m, 1H), 7.49–7.42 (m, 2H), 7.36 (d, $J = 8.2$ Hz, 2H), 7.31–7.24 (m, 3H), 2.80 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.6, 148.1, 147.5, 135.8, 133.5, 131.97, 129.5, 128.9, 128.7, 128.3, 127.3, 126.5, 126.4, 126.22, 126.16, 126.0, 125.8, 125.3, 123.4, 25.5. HRMS (DART-FTICl Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{20}\text{H}_{15}\text{N}$: 270.1277; found: 270.1277.

2-Methyl-4-(thiophen-3-yl)quinoline (3l): Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3l** as a yellow solid (52.2 mg, 77%). Mp 101–102 $^\circ\text{C}$. IR (film, cm^{-1}): 1596; 1414; 1022; 796; 760 cm^{-1} . $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.08 (d, $J = 8.4$ Hz, 1H), 8.00 (d, $J = 8.3$ Hz, 1H), 7.69–7.65 (m, 1H), 7.47–7.41 (m, 3H), 7.30 (dd, $J = 3.5, 2.9$ Hz, 1H), 7.25 (s, 1H), 2.75 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.6, 148.4, 143.2, 138.6, 129.4, 129.1, 128.9, 126.2, 125.9, 125.5, 125.1, 124.8, 122.0, 25.3. HRMS (DART-FTICl Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{14}\text{H}_{11}\text{NS}$: 226.0685; found: 226.0684.

2-Methyl-3,4-diphenylquinoline (3m):¹⁴ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3m** as a yellow solid (46.8 mg, 53%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.12 (d, $J = 8.4$ Hz, 1H), 7.70–7.66 (m, 1H), 7.50 (d, $J = 7.7$ Hz, 1H), 7.38 (t, $J = 7.6$ Hz, 1H), 7.26–7.14 (m, 6H), 7.10–7.05 (m, 4H), 2.55 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 157.9, 147.1, 146.6, 138.7, 136.8, 134.1, 130.13, 130.07, 129.1, 128.7, 127.9, 127.7, 127.2, 126.9, 126.6, 126.3, 125.9, 25.47.

2,3-dimethyl-4-phenylquinoline (3n):¹⁵ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3n** as a yellow solid (50.9 mg, 73%). $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.04 (d, $J = 8.4$ Hz, 1H), 7.59 (dt, $J = 8.4, 4.1$ Hz, 1H), 7.53–7.44 (m, 3H), 7.31 (d, $J = 3.7$ Hz, 2H), 7.24–7.22 (m, 2H), 2.75 (s, 3H), 2.17 (s, 3H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 158.9, 146.3, 146.1, 137.7, 129.4, 128.5, 128.5, 128.2, 127.7, 127.5, 126.9, 126.1, 125.5, 24.6, 17.0.

2-Methyl-4-phenyl-3-(trimethylsilyl)quinoline (3o): Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3o** as a white solid (53.7 mg, 61%). Mp 77–78 $^\circ\text{C}$. IR (film, cm^{-1}): 1562; 1480; 1019; 761; 701 cm^{-1} . $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.97 (d, $J = 8.4$ Hz, 1H), 7.59 (ddd, $J = 8.3, 5.5, 2.7$ Hz, 1H), 7.44–7.42 (m, 3H), 7.29–7.22 (m, 4H), 2.89 (s, 3H), 0.00 (s, 9H). $^{13}\text{C NMR}$ (101 MHz, CDCl_3) δ 160.4, 154.0, 145.0, 137.9, 128.3, 128.2, 127.6, 126.2,

126.02, 125.99, 124.4, 124.0, 123.4, 26.6, 0.0. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{19}H_{21}NSi$: 292.1516; found: 292.1515.

4-Hexyl-2-methylquinoline (3p):¹⁶ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3p** as a white solid (18.1 mg, 27%). 1H NMR (400 MHz, $CDCl_3$) δ 8.01 (dd, $J = 19.8, 8.2$ Hz, 2H), 7.68–7.63 (m, 1H), 7.51–7.47 (m, 1H), 7.13 (s, 1H), 3.03–3.00 (m, 2H), 2.71 (s, 3H), 1.74 (dt, $J = 15.5, 7.6$ Hz, 2H), 1.43 (dd, $J = 14.8, 6.9$ Hz, 2H), 1.37–1.32 (m, 4H), 0.90 (t, $J = 7.0$ Hz, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 158.6, 148.8, 148.0, 129.3, 129.0, 125.9, 125.4, 123.4, 121.6, 32.2, 31.7, 30.1, 29.4, 25.3, 22.6, 14.1.

4-Cyclopropyl-2-methylquinoline (3q):¹² Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3q** as a yellow oil (40.1 mg, 73%). 1H NMR (400 MHz, $CDCl_3$) δ 8.25 (d, $J = 8.2$ Hz, 1H), 8.03 (d, $J = 8.4$ Hz, 1H), 7.68–7.64 (m, 1H), 7.50 (t, $J = 7.3$ Hz, 1H), 6.92 (s, 1H), 2.68 (s, 3H), 2.40–2.33 (m, 1H), 1.14–1.11 (m, 2H), 0.84–0.80 (m, 2H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 158.8, 149.4, 147.7, 129.1, 127.0, 125.4, 123.8, 118.0, 25.4, 12.0, 7.6.

6-(tert-Butyl)-4-(4-methoxyphenyl)-2-methylquinoline (3r). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3r** as a brown solid (68.8 mg, 75%). IR (film, cm^{-1}): 1608; 1461; 1032; 831; 768 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.02 (d, $J = 8.8$ Hz, 1H), 7.87 (s, 1H), 7.78 (d, $J = 8.8$ Hz, 1H), 7.46 (d, $J = 8.4$ Hz, 2H), 7.18 (s, 1H), 7.06 (d, $J = 8.4$ Hz, 2H), 3.99 (m, 3H), 2.74 (s, 3H), 1.34 (s, 9H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 159.7, 157.7, 148.3, 148.1, 146.9, 130.73, 130.67, 128.5, 128.1, 124.7, 122.2, 120.6, 114.0, 55.4, 35.0, 31.2, 25.2. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{21}H_{23}NO$: 306.1852; found: 306.1852.

Methyl 4-(4-Methoxyphenyl)-2-methylquinoline-6-carboxylate (3s). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3s** as a white solid (64.5 mg, 70%). Mp 89–90 °C. IR (film, cm^{-1}): 1716; 1464; 1026; 855; 760 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.66 (d, $J = 1.7$ Hz, 1H), 8.26 (dd, $J = 8.8, 1.8$ Hz, 1H), 8.09 (d, $J = 8.8$ Hz, 1H), 7.45–7.43 (m, 2H), 7.27 (s, 1H), 7.08 (d, $J = 8.6$ Hz, 2H), 3.92 (s, 3H), 3.91 (s, 3H), 2.78 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 166.9, 161.0, 160.1, 150.4, 149.6, 130.8, 129.7, 129.3, 129.0, 128.8, 127.3, 124.5, 122.9, 114.3, 55.4, 52.3, 25.5. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{19}H_{17}NO_3$: 308.1281; found: 308.1280.

4-(4-Methoxyphenyl)-2-methyl-6-(trifluoromethyl)quinoline (3t). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3t** as a yellow solid (32.3 mg, 34%). Mp 61–62 °C. IR (film, cm^{-1}): 1609; 1468; 1031; 832; 780 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.22 (s, 1H), 8.17 (d, $J = 8.8$ Hz, 1H), 7.85 (dd, $J = 8.8, 1.8$ Hz, 1H), 7.43 (d, $J = 8.7$ Hz, 2H), 7.31 (s, 1H), 7.09 (d, $J = 8.7$ Hz, 2H), 3.92 (s, 3H), 2.80 (s, 3H). ^{19}F NMR (377 MHz, $CDCl_3$) δ –62.04. ^{13}C NMR (101 MHz, $CDCl_3$) δ 160.9, 160.2, 149.5, 149.2, 130.7, 130.2, 129.4, 127.5 (q, $J = 32.4$ Hz), 125.0 (q, $J = 3.0$ Hz), 124.5, 124.2 (q, $J = 272.3$ Hz), 123.8 (q, $J = 4.5$ Hz), 123.3, 114.4, 55.4, 25.5. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{18}H_{14}F_3NO$: 318.1100; found: 318.1099.

1-(4-(4-Methoxyphenyl)-2-methylquinolin-6-yl)ethan-1-one (3u). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3u** as a white solid (69.5 mg, 80%). Mp 109–110 °C. IR (film, cm^{-1}): 1675; 1454; 1270; 1021; 880 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.55 (d, $J = 1.8$ Hz, 1H), 8.22 (dd, $J = 8.8, 1.9$ Hz, 1H), 8.09 (d, $J = 8.8$ Hz, 1H), 7.46–7.44 (m, 2H), 7.28 (s, 1H), 7.09–7.07 (m, 2H), 3.91 (s, 3H), 2.78 (s, 3H), 2.60 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 197.6, 161.1, 160.2, 150.5, 149.7, 134.1, 130.8, 129.6, 129.5, 128.0, 127.6, 124.5, 122.9, 114.3, 55.4, 26.7, 25.6. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{19}H_{17}NO_2$: 292.1332; found: 292.1331.

4-(4-Methoxyphenyl)-2-methyl-6-phenylquinoline (3v). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3v** as a brown oil (33.7 mg, 35%). IR (film, cm^{-1}): 1604; 1487; 1245; 1028; 833 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.14 (d, $J = 8.7$ Hz, 1H), 8.09 (d, $J = 2.0$ Hz, 1H), 7.95 (dd, $J = 8.7, 2.1$ Hz, 1H), 7.62–7.60 (m, 2H), 7.49–7.41 (m, 2H), 7.43 (t, $J = 7.5$ Hz, 2H), 7.35–7.32 (m, 1H), 7.23 (s, 1H), 7.06 (d, $J = 8.7$ Hz, 2H), 3.90 (s, 3H), 2.78 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 159.8, 158.5, 148.5, 147.9, 140.8,

138.4, 130.8, 130.4, 129.5, 128.9, 128.9, 127.5, 127.4, 125.4, 123.6, 122.6, 114.1, 55.4, 25.4. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{23}H_{19}NO$: 326.1539; found: 326.1538.

7-Methoxy-4-(4-methoxyphenyl)-2-methylquinoline (3w); 7-Methoxy-4-(4-methoxyphenyl)-2-methylquinoline (3w'). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded as a white solid **3w** (44.4 mg, 53%), **3w'** (20.0 mg, 24%). Mp 96–97 °C. IR (film, cm^{-1}): 1584; 1463; 1029; 878; 834 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 7.78 (d, $J = 9.2$ Hz, 1H), 7.44–7.40 (m, 3H), 7.08 (dd, $J = 9.2, 2.6$ Hz, 1H), 7.08 (s, 1H), 7.03 (d, $J = 8.7$ Hz, 2H), 3.95 (s, 3H), 3.89 (s, 3H), 2.73 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 160.5, 159.8, 158.8, 150.3, 148.2, 130.7, 130.6, 126.8, 120.2, 120.1, 118.5, 114.0, 107.2, 55.5, 55.4, 25.3. HRMS (DART Positive) ($[M + H]^+$) calcd for $C_{18}H_{17}NO_2$: 280.1332; found: 280.1331. **3w'**: Mp 71–72 °C. 1607; 1465; 1032; 872; 833 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 7.70 (dd, $J = 8.5, 0.9$ Hz, 1H), 7.58 (t, $J = 8.1$ Hz, 1H), 7.25 (d, $J = 8.8$ Hz, 2H), 7.06 (s, 1H), 6.94–6.91 (m, 2H), 6.78 (d, $J = 7.7$ Hz, 1H), 3.88 (s, 3H), 3.55 (s, 3H), 2.72 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 158.7, 158.1, 156.5, 150.0, 147.8, 135.1, 129.5, 129.4, 124.3, 121.6, 117.3, 112.4, 105.6, 55.4, 55.3, 24.9. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{18}H_{17}NO_2$: 280.1332; found: 280.1331.

4-(4-Methoxyphenyl)-2-methylbenzo[h]quinoline (3x). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3x** as a white solid (51.8 mg, 58%). Mp 111–112 °C. IR (film, cm^{-1}): 1589; 1497; 1033; 837; 777 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 9.40 (d, $J = 8.1$ Hz, 1H), 7.84 (d, $J = 7.1$ Hz, 1H), 7.78 (d, $J = 9.1$ Hz, 1H), 7.72–7.62 (m, 3H), 7.42 (d, $J = 8.7$ Hz, 2H), 7.29 (s, 1H), 7.03 (d, $J = 8.7$ Hz, 2H), 3.86 (s, 3H), 2.83 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 159.7, 157.1, 148.1, 146.5, 133.5, 131.6, 130.9, 128.0, 127.5, 126.8, 126.4, 124.9, 123.1, 122.7, 122.5, 114.0, 55.4, 25.4. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{21}H_{17}NO$: 300.1383; found: 300.1382.

4-(4-Methoxyphenyl)-2-phenylquinoline (3y):¹⁷ Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3y** as a white solid (50.3 mg, 54%). 1H NMR (400 MHz, $CDCl_3$) δ 8.23 (d, $J = 8.0$ Hz, 1H), 8.20–8.18 (m, 2H), 7.95 (dd, $J = 8.4, 0.8$ Hz, 1H), 7.79 (s, 1H), 7.72 (ddd, $J = 8.3, 6.9, 1.4$ Hz, 1H), 7.54–7.44 (m, 6H), 7.08 (d, $J = 8.7$ Hz, 2H), 3.90 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 159.9, 157.0, 148.9, 139.8, 130.9, 130.7, 130.2, 129.5, 129.3, 128.9, 127.6, 126.3, 126.0, 125.7, 119.4, 114.1, 55.5.

2-(tert-Butyl)-4-(4-methoxyphenyl)quinoline (3z). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3z** as a yellow oil (45.4 mg, 52%). IR (film, cm^{-1}): 1609; 1496; 1245; 1033; 832 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.11 (d, $J = 8.4$ Hz, 1H), 7.88 (d, $J = 8.1$ Hz, 1H), 7.67–7.63 (m, 1H), 7.46–7.39 (m, 4H), 7.05 (d, $J = 8.6$ Hz, 2H), 3.88 (s, 3H), 1.49 (s, 9H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 168.8, 159.7, 148.1, 147.9, 131.2, 130.9, 129.8, 128.9, 125.6, 125.5, 125.3, 118.5, 114.0, 55.4, 38.2, 30.2. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{20}H_{21}NO$: 292.1696; found: 292.1695.

Methyl 4-(4-(Methoxycarbonyl)phenyl)-2-methylquinoline-6-carboxylate (3aa). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3aa** as a white solid (35.5 mg, 35%). Mp 188–189 °C. IR (film, cm^{-1}): 1715; 1563; 1277; 1103; 857 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.52 (d, $J = 1.5$ Hz, 1H), 8.29 (dd, $J = 8.8, 1.7$ Hz, 1H), 8.23 (d, $J = 8.2$ Hz, 2H), 8.13 (d, $J = 8.8$ Hz, 1H), 7.59 (d, $J = 8.2$ Hz, 2H), 7.31 (s, 1H), 4.00 (s, 3H), 3.91 (s, 3H), 2.81 (s, 3H). ^{13}C NMR (101 MHz, $CDCl_3$) δ 166.7, 161.0, 150.3, 148.7, 142.0, 130.5, 130.0, 129.6, 129.5, 129.2, 128.4, 127.6, 123.9, 122.9, 52.37, 52.36, 25.6. HRMS (DART-FTICl Positive) ($[M + H]^+$) calcd for $C_{20}H_{17}NO_4$: 336.1230; found: 336.1231.

Methyl 4-(3-((Allyloxy)carbonyl)amino)phenyl)-2-methylquinoline-6-carboxylate (3ab). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3ab** as a white solid (52.5 mg, 47%). Mp 83–84 °C. IR (film, cm^{-1}): 1721; 1593; 1278; 1050; 845 cm^{-1} . 1H NMR (400 MHz, $CDCl_3$) δ 8.60 (d, $J = 1.7$ Hz, 1H), 8.26 (dd, $J = 8.8, 1.9$ Hz, 1H), 8.10 (d, $J = 8.8$ Hz, 1H), 7.59 (s, 1H), 7.53 (d, $J = 8.4$ Hz, 1H), 7.47 (t, $J = 7.8$ Hz, 1H), 7.29 (s, 1H), 7.20–7.17 (m, 1H), 7.06 (s, 1H), 5.96 (ddt, $J = 16.1, 10.5, 5.7$ Hz,

1H), 5.36 (ddd, $J = 17.2, 2.9, 1.4$ Hz, 1H), 5.26 (dd, $J = 10.4, 1.2$ Hz, 1H), 4.68 (d, $J = 5.7$ Hz, 2H), 3.91 (s, 3H), 2.78 (s, 3H). ^{13}C NMR (101 MHz, CDCl_3) δ 166.9, 161.0, 153.2, 150.3, 149.4, 138.40, 138.38, 132.3, 129.5, 129.4, 129.0, 128.9, 127.39, 124.7, 124.3, 123.0, 119.5, 118.9, 118.5, 66.0, 52.3, 25.6. HRMS (DART-FTIC Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{22}\text{H}_{20}\text{N}_2\text{O}_4$: 377.1496; found: 377.1496.

1-(4-Cyclopropyl-2-methylquinolin-6-yl)ethan-1-one (**3ac**). Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **3ac** as a white solid (41.1 mg, 61%). Mp 88–89 °C. IR (film, cm^{-1}): 1670; 1596; 1257; 1063; 831 cm^{-1} . ^1H NMR (400 MHz, CDCl_3) δ 8.92 (d, $J = 1.8$ Hz, 1H), 8.21 (dd, $J = 8.8, 1.9$ Hz, 1H), 8.05 (d, $J = 8.8$ Hz, 1H), 6.97 (s, 1H), 2.75 (s, 3H), 2.71 (s, 3H), 2.48 (ddd, $J = 13.7, 8.3, 5.4$ Hz, 1H), 1.23 (ddd, $J = 8.4, 6.3, 4.5$ Hz, 2H), 0.91–0.87 (m, 2H). ^{13}C NMR (101 MHz, CDCl_3) δ 197.8, 161.5, 151.3, 149.8, 133.8, 129.6, 127.6, 126.3, 125.7, 118.58, 26.8, 25.6, 11.9, 8.3. HRMS (DART-FTIC Positive) ($[\text{M} + \text{H}]^+$) calcd for $\text{C}_{15}\text{H}_{15}\text{NO}$: 226.1226; found: 226.1226.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.6b02509.

Characterization of products (copies of ^1H , ^{13}C , and ^{19}F NMR spectra) (PDF)

■ AUTHOR INFORMATION

Corresponding Authors

*E-mail: yushouyun@nju.edu.cn (S.Y.).

*E-mail: txdqxd1979@163.com (S.S.).

ORCID

Shouyun Yu: 0000-0003-4292-4714

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

Financial support from the National Natural Science Foundation of China (21472084 and 81421091) and the State Key Laboratory of Bioorganic Chemistry, Shanghai Institute of Organic Chemistry, CAS is acknowledged.

■ REFERENCES

- (1) (a) Sanders, J. M.; Clifford, D. P.; Lutz, R. E. *J. Med. Chem.* **1971**, *14*, 1126. (b) Natarajan, J. K.; Alumasa, J. N.; Yearick, K.; Ekoue-Kovi, K. A.; Casabianca, L. B.; de Dios, A. C.; Wolf, C.; Roepe, P. D. *J. Med. Chem.* **2008**, *51*, 3466. (c) Lord, A.-M.; Mahon, M. F.; Lloyd, M. D.; Threadgill, M. D. *J. Med. Chem.* **2009**, *52*, 868. (d) Bax, B. D.; Chan, P. F.; Eggleston, D. S.; Fosberry, A.; Gentry, D. R.; Gorrec, F.; Giordano, I.; Hann, M. M.; Hennessy, A.; Hibbs, M.; Huang, J.; Jones, E.; Jones, J.; Brown, K. K.; Lewis, C. J.; May, E. W.; Saunders, M. R.; Singh, O.; Spitzfaden, C. E.; Shen, C.; Shillings, A.; Theobald, A. J.; Wohlkonig, A.; Pearson, N. D.; Gwynn, M. N. *Nature* **2010**, *466*, 935. (e) Andrews, S.; Burgess, S. J.; Skaalrud, D.; Kelly, J. X.; Peyton, D. H. *J. Med. Chem.* **2010**, *53*, 916. (f) Mukherjee, S.; Pal, M. *Drug Discovery Today* **2013**, *18*, 389. (g) Afzal, O.; Kumar, S.; Haider, M. R.; Ali, M. R.; Kumar, R.; Jaggi, M.; Bawa, S. *Eur. J. Med. Chem.* **2015**, *97*, 871.
- (2) (a) Li, H.; Jäkke, F. *Macromolecules* **2009**, *42*, 3448. (b) Tao, S.; Li, L.; Yu, J.; Jiang, Y.; Zhou, Y.; Lee, C.-S.; Lee, S.-T.; Zhang, X.; Kwon, O. *Chem. Mater.* **2009**, *21*, 1284. (c) Artizzu, F.; Mercuri, M. L.; Serpe, A.; Deplano, P. *Coord. Chem. Rev.* **2011**, *255*, 2514. (d) Kim, J. I.; Shin, I.-S.; Kim, H.; Lee, J.-K. *J. Am. Chem. Soc.* **2005**, *127*, 1614.
- (3) (a) Tan, B.; Shi, Z.; Chua, P. J.; Zhong, G. *Org. Lett.* **2008**, *10*, 3425. (b) Cheng, S.; Yu, S. *Org. Biomol. Chem.* **2014**, *12*, 8607. (c) Cheng, S.; Zhao, L.; Yu, S. *Adv. Synth. Catal.* **2014**, *356*, 982. (d) Biddle, M. M.; Lin, M.; Scheidt, K. A. *J. Am. Chem. Soc.* **2007**, *129*, 3830. (e) Chen, G.; Gong, W.; Zhuang, Z.; Andr , M. S.; Chen, Y.-Q.

Hong, X.; Yang, Y.-F.; Liu, T.; Houk, K. N.; Yu, J.-Q. *Science* **2016**, *353*, 1023. (f) Delapierre, G.; Brunel, J. M.; Constantieux, T.; Buono, G. *Tetrahedron: Asymmetry* **2001**, *12*, 1345. (g) Zhang, Y.; Sigman, M. S. *J. Am. Chem. Soc.* **2007**, *129*, 3076. (h) Michel, B. W.; Steffens, L. D.; Sigman, M. S. *J. Am. Chem. Soc.* **2011**, *133*, 8317. (i) Abrunhosa, L.; Delain-Bioton, L.; Gaumont, A.-C.; Gulea, M.; Masson, S. *Tetrahedron* **2004**, *60*, 9263.

(4) For selected reviews: (a) Madapa, S.; Tusi, Z.; Batra, S. *Curr. Org. Chem.* **2008**, *12*, 1116. (b) Khusnutdinov, R. I.; Bayguzina, A. R.; Dzhemilev, U. M. *J. Organomet. Chem.* **2014**, *768*, 75. (c) Bharate, J. B.; Vishwakarma, R. A.; Bharate, S. B. *RSC Adv.* **2015**, *5*, 42020. (d) Batista, V. F.; Pinto, D. C. G. A.; Silva, A. M. S. *ACS Sustainable Chem. Eng.* **2016**, *4*, 4064. (e) Ramann, G.; Cowen, B. *Molecules* **2016**, *21*, 986.

(5) (a) Marco-Contelles, J.; P rez-Mayoral, E.; Samadi, A.; Carreiras, M. d. C.; Soriano, E. *Chem. Rev.* **2009**, *109*, 2652. (b) Henze, H. R.; Carroll, D. W. *J. Am. Chem. Soc.* **1954**, *76*, 4580.

(6) For selected recent examples: (a) Wang, Y.; Chen, C.; Peng, J.; Li, M. *Angew. Chem., Int. Ed.* **2013**, *52*, 5323. (b) Wang, X.; Wang, X.; Huang, D.; Liu, C.; Wang, X.; Hu, Y. *Adv. Synth. Catal.* **2016**, *358*, 2332. (c) Aradi, K.; Bombicz, P.; Nov k, Z. *J. Org. Chem.* **2016**, *81*, 920. (d) Yi, X.; Xi, C. *Org. Lett.* **2015**, *17*, 5836. (e) Yan, Q.; Chen, Z.; Liu, Z.; Zhang, Y. *Org. Chem. Front.* **2016**, *3*, 678. (f) Jiang, H.; An, X.; Tong, K.; Zheng, T.; Zhang, Y.; Yu, S. *Angew. Chem., Int. Ed.* **2015**, *54*, 4055. (g) Zhu, R.; Cheng, G.; Jia, C.; Xue, L.; Cui, X. *J. Org. Chem.* **2016**, *81*, 7539. (h) Yao, R.; Rong, G.; Yan, B.; Qiu, L.; Xu, X. *ACS Catal.* **2016**, *6*, 1024. (i) Zheng, J.; Li, Z.; Huang, L.; Wu, W.; Li, J.; Jiang, H. *Org. Lett.* **2016**, *18*, 3514.

(7) For recent selected examples: (a) Bharate, J. B.; Wani, A.; Sharma, S.; Reja, S. I.; Kumar, M.; Vishwakarma, R. A.; Kumar, A.; Bharate, S. B. *Org. Biomol. Chem.* **2014**, *12*, 6267. (b) Majumdar, K. C.; Ponra, S.; Ghosh, D.; Taher, A. *Synlett* **2011**, *2011*, 104. (c) Zhang, X.; Xu, X.; Yu, L.; Zhao, Q. *Asian J. Org. Chem.* **2014**, *3*, 281.

(8) Al-Talib, M.; Jochims, J. C.; Wang, Q.; Hamed, A.; Ismail, A. E.-H. *Synthesis* **1992**, *1992*, 875.

(9) For related reviews: (a) Galli, C. *Chem. Rev.* **1988**, *88*, 765. (b) Roglans, A.; Pla-Quintana, A.; Moreno-Ma nas, M. *Chem. Rev.* **2006**, *106*, 4622. (c) Bonin, H.; Fouquet, E.; Felpin, F.-X. *Adv. Synth. Catal.* **2011**, *353*, 3063. (d) Hari, D. P.; K nig, B. *Angew. Chem., Int. Ed.* **2013**, *52*, 4734. (e) Oger, N.; Le Grogne, E.; Felpin, F.-X. *Org. Chem. Front.* **2015**, *2*, 590. Selected recent works on the heterocycles synthesis involving nitriles and alkynes: (f) Xie, L. G.; Niyomchon, S.; Mota, A. J.; Gonzalez, L.; Maulide, N. *Nat. Commun.* **2016**, *7*, 10914. (g) Wang, Y.; Song, L. J.; Zhang, X.; Sun, J. *Angew. Chem., Int. Ed.* **2016**, *55*, 9704. (h) Zhang, J.; Zhang, Q.; Xia, B.; Wu, J.; Wang, X. N.; Chang, J. *Org. Lett.* **2016**, *18*, 3390.

(10) (a) Xia, Z.; Zhu, Q. *Org. Lett.* **2013**, *15*, 4110. (b) Ramanathan, M.; Liu, S.-T. *J. Org. Chem.* **2015**, *80*, 5329.

(11) Hanson, P.; Jones, J. R.; Taylor, A. B.; Walton, P. H.; Timms, A. W. *J. Chem. Soc., Perkin Trans. 2* **2002**, 1135.

(12) Li, W.; Gao, J. J.; Zhang, Y.; Tang, W.; Lee, H.; Fandrick, K. R.; Lu, B.; Senanayake, C. H. *Adv. Synth. Catal.* **2011**, *353*, 1671.

(13) Takahashi, K.; Mitsuhashi, K. *J. Heterocycl. Chem.* **1977**, *14*, 881.

(14) Kong, L.; Yu, S.; Zhou, X.; Li, X. *Org. Lett.* **2016**, *18*, 588.

(15) Zhao, P.; Yan, X.; Yin, H.; Xi, C. *Org. Lett.* **2014**, *16*, 1120.

(16) Fontana, F.; Minisci, F.; Nogueira Barbosa, M.; Vismara, E. *Acta Chem. Scand.* **1989**, *43*, 995.

(17) Akiyama, S.; Nakashima, S.; Yokota, K.; Fuchibe, K. *Chem. Lett.* **2004**, *33*, 922.